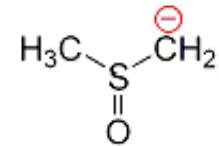
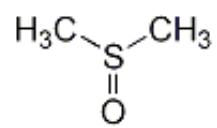
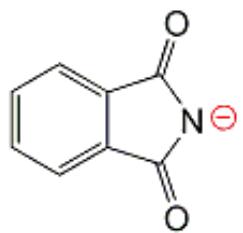
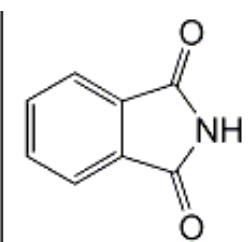


Forme acide	Forme basique	pKa	Forme acide	Forme basique	pKa
HClO ₄	ClO ₄ ⁻	-10			9
HI	I ⁻	-10			9
	R-C≡N	-10	HCN	CN ⁻	9.2
		-10	NH ₄ ⁺	NH ₃	9.24
H ₂ SO ₄	HSO ₄	-10			8-11
HBr	Br ⁻	-9			10
HCl	Cl ⁻	-7	RNH ₃ ⁺	RNH ₂	10-11
		-7	HCO ₃ ⁻	CO ₃ ²⁻	10.33
ArSO ₃ H	ArSO ₃ ⁻	-6.5			10.66
		-6.5	RSH	RS ⁻	10-11
ArOH ₂ ⁺	ArOH	-6.4	NC-CH ₂ -CN		11
		-6			11
		-5	HOOH	HOO ⁻	11.6

Ar_3NH^+	Ar_3N	-5			12.5
	$\text{H}-\text{C}(=\text{O})-\text{H}$	-4			13
	$\text{R}-\text{O}-\text{R}'$	-3.5			13.5
ROH_2^+	ROH	-2	CH_3OH	CH_3O^-	15.2
H_3O^+	H_2O	-1.74	H_2O	OH^-	15.74
HNO_3	NO_3^-	-1.4			16
	$\text{R}-\text{C}(=\text{O})-\text{NH}_2$	-0.5	$\text{R}-\text{CH}_2-\text{OH}$	$\text{R}-\text{CH}_2-\text{O}^-$	16
Ar_2NH_2^+	Ar_2NH	1			17
HF	F^-	3.17			19-20
ArNH_3^+	ArNH_2	3-5			24.5
		4-5			25
		5	$\text{H}-\text{C}\equiv\text{C}-\text{H}$	$\text{H}-\text{C}\equiv\text{C}^-$	25
H_2S	HS^-	7			25
		8.30			31.5



33

 H_2 H^-

35

 $i\text{Pr}_2\text{NH}$ $i\text{Pr}_2\text{N}^-$

36

 NH_3 NH_2^-

38

 RH R^-

50